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# Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

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		Applic	cation No. Applicant(s)				
		10/581	,446	ORIGUCHI ET AL.			
Office Action Summary			ner	Art Unit			
		MICHA	EL A. SALVITTI	1796			
Period fo	The MAILING DATE of this communic or Reply	ation appears on	the cover sheet with the	correspondence ac	ddress		
A SHO WHIC - Exter after - If NO - Failur Any r	ORTENED STATUTORY PERIOD FOR CHEVER IS LONGER, FROM THE MA asions of time may be available under the provisions of SIX (6) MONTHS from the mailing date of this community period for reply is specified above, the maximum stature to reply within the set or extended period for reply within	ILING DATE OF 37 CFR 1.136(a). In no ication. tory period will apply an II, by statute, cause the	THIS COMMUNICATION event, however, may a reply be to divide white spire SIX (6) MONTHS from application to become ABANDON	N. imely filed in the mailing date of this of ED (35 U.S.C. § 133).	•		
Status							
2a)⊠	Responsive to communication(s) filed This action is <b>FINAL</b> . 2b Since this application is in condition for closed in accordance with the practice	)∏ This action i r allowance exce	 s non-final. ept for formal matters, pr		e merits is		
Dispositi	on of Claims		•				
5)□ 6)⊠ 7)□ 8)□ Applicati	Claim(s) 1-17 is/are pending in the ap 4a) Of the above claim(s) is/are Claim(s) is/are allowed. Claim(s) 1-17 is/are rejected. Claim(s) is/are objected to. Claim(s) are subject to restriction on Papers The specification is objected to by the	withdrawn from					
_	The drawing(s) filed on is/are: a Applicant may not request that any objecti Replacement drawing sheet(s) including the The oath or declaration is objected to be	on to the drawing( ne correction is rec	s) be held in abeyance. Se uired if the drawing(s) is ol	ee 37 CFR 1.85(a). ojected to. See 37 C	, ,		
Priority u	ınder 35 U.S.C. § 119						
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  a) All b) Some * c) None of:  1. Certified copies of the priority documents have been received.  2. Certified copies of the priority documents have been received in Application No  3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).  * See the attached detailed Office action for a list of the certified copies not received.							
2)  Notic 3) Inforr	t(s) e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTO nation Disclosure Statement(s) (PTO/SB/08) r No(s)/Mail Date	D-948)	4) Interview Summar Paper No(s)/Mail D 5) Notice of Informal 6) Other:	Date			

# **DETAILED ACTION**

#### Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-10 and 17 are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 5,320,769 to *Botnick et al.* in view of U.S. Patent No. 5,840,800 to *Joffre et al.* 

Regarding claim 1: Botnick teaches a vinyl-urethane copolymer comprising at least one vinyl polymer chain and at least one urethane polymer chain (Botnick col. 2, lines 57-63). The two polymer chains are combined through an intermediary linkage segment (self-crosslinking; Botnick col. 2, line 64). This linkage segment is bonded to the vinyl chain and the urethane chain.

Botnick does not teach the linkage segment comprising a silicon oxygen bond.

Joffre teaches organic polymers comprising silicon modified organic polymers (Joffre col. 4, lines 42-46). These polymers crosslink to form silicon-oxygen bonds between chains on the polymers (Joffre col. 9, lines 25-30). Botnick and Joffre are analogous art in that they are drawn to the same field of endeavor, namely aqueous dispersions of self-crosslinking vinyl polymers. At the time of the invention, it would have been obvious to a person having ordinary skill in the art to substitute the reactive self-

crosslinking sites in the polymers of *Botnick* with the reactive silyl groups (as taught by *Joffre*). The motivation in doing so is the creation of compositions showing elasticity upon drying (*Joffre* col. 1, lines 35-40) that have advantageous properties regarding shelf-life and toxicity (*Joffre* col. 3, lines 16-20).

Regarding claim 2: Botnick teaches the urethane polymer is a residue of a urethane polymer (A) having at least one reactive group (free hydroxyl groups of Example 1 - analogous to the silicon-containing group for the purposes of bonding the vinyl and urethane chains). The functionalized polyurethane polymer resins are mixed with functionalized vinyl resins (see Botnick col. 19, lines 35-53). The functionalized vinyl polymer is derived from an ethylenically unsaturated monomer (B) (methyl methacrylate) and a compound (C) (acetoacetoxyethyl methacrylate; see Example 11, col. 17, lines 5-20). Compound (C) is analogous to the silicon-containing monomer in the instant recited claim, in that the acetoacetoxyethyl undergoes crosslinking reactions with free anionic salt groups (deprotonated hydroxyls) on the functionalized polyurethane (Botnick col. 3, lines 14-22).

Regarding claims 3, 4 and 8: Botnick in view of Joffre teaches the composition of claim 1, as set forth above. Botnick teaches a crosslinking reaction occurring between acetoacetoxyethyl groups on the vinyl polymer with the free anionic salt groups on the functionalized polyurethane (Botnick col. 3, lines 14-22).

Botnick is silent regarding the linkage segment between the polymers as being silicon polymer chains. Joffre teaches crosslinking polymer chains with hydrolysable silicone moieties (Joffre col. 4, line 65 through col. 5, line ) wherein crosslinking

between the polymer chains is achieved through condensation of the hydrolysable silicone moieties (*Joffre* col. 14, lines 12-17). *Joffre* teaches silane compounds (D) as hydrolysable alkoxysilylated groups (see claim 15). *Botnick* and *Joffre* are analogous art in that they are drawn to the same field of endeavor, namely aqueous dispersions of self-crosslinking vinyl polymers. At the time of the invention, it would have been obvious to a person having ordinary skill in the art to substitute the reactive self-crosslinking sites in the polymers of *Botnick* with the reactive silyl groups (as taught by *Joffre*). The motivation in doing so is the creation of compositions showing elasticity upon drying (*Joffre* col. 1, lines 35-40) that have advantageous properties regarding shelf-life and toxicity (*Joffre* col. 3, lines 16-20).

Regarding claims 5-7: Botnick in view of Joffre teaches the composition of claim 2, as set forth above. Botnick teaches the urethane polymer (A) having the reactive group (analogous to the silicon group) further contains at least one monomer with a hydrophilic group (trimetholol propane; col. 12, line 10). The functional-modified urethane polymer (A1) contains a hydrophilic group (polyol from the Tone 210 Polyol). The (A1) urethane polymer comprises 2-methylpentamethylenediamine (A1-a), caprolactone diol (A1-b), 1,3-bis(1-isocyanato-1-methylethyl)benzene (A1-c), trimethylolpropane (A1-d, given broadest reasonable interpretation as being equivalent to the alkoxysilyl since this the free hydroxyl crosslinks the polymers).

Botnick is silent regarding the reactive groups containing alkoxysilation. Joffre teaches silane compounds as hydrolysable alkoxysilylated groups (see claim 15). At the time of the invention, it would have been obvious to a person having ordinary skill in the

art to incorporate hydrolysable alkoxysilyl groups as the crosslinking units in the polymers of *Botnick*, with the motivation of enabling crosslinking without the need of a crosslinker (*Joffre*, col. 9, lines 25-30), allowing for omission of the crosslinking agent and therefore the composition can be made at lower cost.

Botnick is silent regarding embodiments wherein the reactive groups are at the terminus of the polymer chains. Joffre teaches that reactive groups at the termini are desirable (col. 4, lines 49-53). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to incorporate reactive silyl groups into the termini of the functional polymers taught by Botnick, with the motivation of increasing the strength and elongation of the cured product (Joffre, col. 4, lines 49-53).

Regarding claims 9 and 10: Botnick teaches the vinyl as an acrylic (methyl methacrylate; Example 11, col. 17). The compound C having the reactive group (acetoacetoxyethyl methacrylate) also contains an ethylenically unsaturated bond (Example 11, col. 17). As above, the acetoacetoxyethyl methacrylate is responsible for crosslinking the polymers in this composition (Botnick col. 3, lines 14-22) and it has been examined as being an equivalent to the silyls of the instant claims, in terms of monomers and presence in the respective polymers.

Regarding claim 17: Botnick teaches the vinyl-urethane copolymer as an aqueous dispersion (col. 19, lines 35-53).

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Claims 11-16 are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 5,320,769 to *Botnick et al.* in view of U.S. Patent No. 5,840,800 to *Joffre et al.* 

Regarding claim 11: Botnick teaches a method for producing vinyl-urethane copolymer comprising at least one vinyl polymer chain and at least one urethane polymer chain (Botnick col. 2, lines 57-63). The two polymer chains are combined through an intermediary linkage segment (self-crosslinking; Botnick col. 2, line 64). This linkage segment is bonded to the vinyl chain and the urethane chain.

As to the method, *Botnick* teaches Step (X) wherein the urethane polymer with the reactive group is prepared (Example 1, col. 12), followed by (Y) polymerization of compounds B and C (methyl methacrylate and acetoacetoxyethyl methacrylate, respectively; Example 9, col. 16) <u>during the polymerization reaction</u> in the presence of the functionalized urethane (A) (see Example 35, col. 19).

Botnick does not teach the linkage segment comprising a silicon oxygen bond.

Joffre teaches organic polymers comprising silicon modified organic polymers (Joffre col. 4, lines 42-46). These polymers crosslink to form silicon-oxygen bonds between chains on the polymers (Joffre col. 9, lines 25-30). Botnick and Joffre are analogous art in that they are drawn to the same field of endeavor, namely aqueous dispersions of self-crosslinking vinyl polymers. At the time of the invention, it would have been obvious to a person having ordinary skill in the art to substitute the reactive self-crosslinking sites in the polymers of Botnick with the reactive silyl groups (as taught by Joffre). The motivation in doing so is the creation of compositions showing elasticity

upon drying (*Joffre* col. 1, lines 35-40) that have advantageous properties regarding shelf-life and toxicity (*Joffre* col. 3, lines 16-20).

Regarding claim 12: Botnick in view of Joffre teaches the method of claim 11, as set forth above.

Botnick teaches Step (Y1-a) wherein a dispersion of functionalized polyurethane (A) is prepared, followed by simultaneously polymerizing B (methyl methacrylate) and C (acetoacetoxy ethylmethacrylate); see Example 35, col. 19).

Regarding claim 13: Botnick in view of Joffre teaches the method of claim 11, as set forth above.

Botnick teaches Step (Y2-C) wherein a dispersion of functionalized polyurethane (A) is prepared, followed by polymerizing the vinyl emulsion of Example 9 (Example 35, col. 19). Monomers B and C are polymerized simultaneously <u>during</u> the polymerization reaction; the equivalent of the condensation reaction (crosslinking vinyl and urethane) occurs <u>before</u> the <u>condensation</u> (curing time is days (see Table VII, col. 26).

Botnick is silent regarding hydrolysis of a silane compound (D) in this reaction.

Joffre teaches crosslinking polymer chains with hydrolysable silicone moieties (Joffre col. 4, line 65 through col. 5, line) wherein crosslinking between the polymer chains is achieved through condensation of the hydrolysable silicone moieties (Joffre col. 14, lines 12-17). Joffre teaches silane compounds (D) as hydrolysable alkoxysilylated groups (see claim 15). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to substitute the reactive self-crosslinking sites in the polymers of Botnick with the reactive silyl groups (as taught by Joffre). The

motivation in doing so is the creation of compositions showing elasticity upon drying (*Joffre* col. 1, lines 35-40) that have advantageous properties regarding shelf-life and toxicity (*Joffre* col. 3, lines 16-20).

Regarding claim 14: Botnick teaches the reactive functional group compound (acetoxyethyl methacrylate in Example 9; analogous to the silane compound D) has at least one functional group reactive with an ethylenically unsaturated bond-containing group in combination with a silane compound free from a functional group reactive with an ethylenically unsaturated monomer (methyl methacrylate in Example 9).

Regarding claim 15: Botnick teaches an embodiment wherein ethylenically unsaturated monomer acts as a solvent during the preparation of the urethane polymer (Example 36, col. 20). This has been interpreted on the basis that some free monomer is present in Example 9 (col. 16, lines 56-57).

Regarding claim 16: Botnick teaches a method wherein the urethane polymer (A) having at least one silicon-containing hydrolysable group used in step (X) is a hydroxy-functionalized (equivalent to alkoxysilane) urethane polymer (A1) containing at least one hydrophilic group (Example 1 has both a hydrophilic group from polycaprolactone diol, and a free hydroxyl from trimethylolpropane).

# Response to Arguments

The following responses are directed to the document entitled "Remarks" submitted on August 17<sup>th</sup>, 2009.

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**A)** Applicant's arguments with respect to the rejection(s) of claim(s) 1 and 3 to Lai (Journal of Applied Polymer Science, 56, 301-310 (1995)) under 35 U.S.C. 102(b) have been fully considered and are persuasive. Therefore, the rejection has been withdrawn. However, upon further consideration, a new ground(s) of rejection is made in view of U.S. Patent No. 5,541,251 to Botnick and U.S. Patent No. 5,840,800 to Joffre.

- **B)** Applicant's arguments with respect to claims 1-10 to WO/0301865 to Schindler under 102(b) have been considered but are moot in view of the new ground(s) of rejection.
- C) Applicant's arguments with respect to claims 11-16 to WO/0301865 (*Schindler*), U.S. Patent No. 6,031,041 (*Chung*) and U.S. Patent No. 5,854,332 (*Swarup*) under 103(a) have been considered but are moot in view of the new ground(s) of rejection.

#### Conclusion

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the

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shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

### Correspondence

Any inquiry concerning this communication or earlier communications from the examiner should be directed to MICHAEL A. SALVITTI whose telephone number is (571)270-7341. The examiner can normally be reached on Monday-Thursday 8AM-7PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark Eashoo can be reached on (571) 272-1197. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Mark Eashoo/ /M. A. S./

Supervisory Patent Examiner, Art Unit 1796 Examiner, Art Unit 1796